

REMARKS/ARGUMENTS

Claim 1-31 are pending and are amended herein. These claims have been amended to remove multiple dependencies and present the claims into better compliance with standard U.S. claim format. No new matter has been added by these amendments.

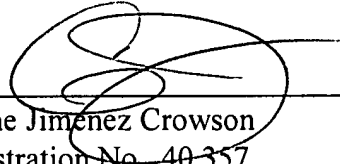
Entry of the above amendments is respectfully requested.

Respectfully submitted,

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DESCRIPTION

~~"A-CHEMICAL COMPOSITION AND THE METHOD OF POLYMERISATION
THEREOF FOR APPLICATIONS USE WITHIN THE COACHBUILDING
SECTOR ON VEHICLE BODYWORK REPAIR BODY"~~

5 FIELD OF INVENTION

[0001]. The present invention relates to a chemical composition and a polymerisation method for said composition, and relates in particular to a composition and a method useful in the vehicle bodywork repair sector.

10

BACKGROUND OF THE INVENTION

[0002]. ~~Currently,~~ Vehicle bodywork repairs are currently carried out in various stages which usually comprise, in sequence, the reshaping or the substitution
15 replacement of the damaged parts, the filling and later sandpapering, the application of one or more primer(s) and relevant polymerisation(s), the preparation of the surfaces for varnishing using abrasive papers, the application of a "base coat" and its the drying thereof
20 at room temperature, the application of a "clear coat 2K" and its the polymerisation thereof through a cooking cycle in an oven at 60°C for 30-60' or at room temperature for approx. 24 hours. Lastly
25 any possible defects arising from the previous stages

and, in the case of work on limited areas of bodywork (retouching), polishing with abrasive (cutting) paste and polish in order to blend in the repaired and revarnished parts with these not affected by the retouching.

5 [0003]. In particular, by the term "~~foundation~~primer" is meant any type of varnish used in the sector which has the function of preparing the metal bodywork panel for the adhesion of new original varnish to said bodywork panel, and for ~~its~~the protection thereof.

10 [0004]. By the term "base coat" is meant a metallicised, -micalized or coloured varnish having a colouring function which restores the colour of the original bodywork where it has been damaged.

[0005]. By the term "clear coat 2K" is meant a
15 transparent varnish ~~constituted by~~consisting of a polymerisable component and ~~by~~a catalyst essential for said polymerisation.

[0006]. Bodywork repairs for vehicles, or parts ~~of~~
~~them~~thereof, according to the above-mentioned procedure
20 ~~have~~has some inconveniences. Firstly, the polymerisation execution times for the oven cooking of the "clear coat 2K" are particularly long. Furthermore, prior to being able to carry out the final stage of the elimination of any possible defects, ~~it is necessary that~~the "clear
25 coat 2K" requires to be perfectly polymerised otherwise

it is impossible to carry out the polishing of the treated area. Polymerisation by heating in an oven however, is not usually prolonged for more than 20-35' due to the high costs of the fuels used, whereby, the degree of polymerisation obtained in this stage is never complete. This implies however that the treated part must be left in order to complete polymerisation for even 24 hours, with a significant delay for the finishing stage. It is ~~clear~~ understood that these times have repercussions on the delivery of the repaired vehicle with high costs which must be supported by the owner.

SUMMARY OF THE INVENTION

[0007]. The problem at the heart of the present invention is therefore that of ~~devising~~ providing a vehicle bodywork repair method which is rapid, reliable and economically advantageous, both for the repairer and for the owner of the repaired vehicle.

[0008]. ~~Such a~~ This problem is resolved by a method as covered by the attached claims.

20 [0009]. Accordingly, the first aim of the present invention ~~has therefore, as first subject, is the making to provide available of~~ a transparent polymerisable chemical composition which is particularly useful in the vehicle bodywork repair sector.

25 [0010]. A second ~~subject~~ aim of the invention is a

~~procedure~~process for the preparation of said chemical composition.

[0011]. An further aim~~additional subject~~ is a polymerisation method for said chemical composition.

5 [0012]. A still further~~nether additional subject~~aim is the use of a polymerisable chemical composition for the repair of the bodywork of vehicles or parts ~~of them~~thereof.

[0013]. Further characteristics and the advantages of
10 the present invention will ~~become clearer~~be more understood from the following description of an example embodiment given for non limiting indicative purposes.

DETAILED DESCRIPTION OF THE INVENTION

[0014]. The transparent polymerisable chemical
15 composition of the invention can be advantageously used for applications in the sector of repairs to the bodywork of vehicles or parts ~~of them~~thereof. ~~Such a~~This composition comprises ~~from~~10% to 60% of transparent hydroxylated acrylic resins, ~~from~~10% to 70% of
20 oligoethers and monomers selected from acrylates or methacrylates, ~~from~~0% to 90% of solvents and ~~from~~0.1% to 10 % of photoinitiators which initiate the polymerisation process when stimulated by light radiation.

25 [0015]. In particular, the composition comprises ~~from~~

20% to 50% of resin, ~~from~~ 20% to 70% of monomers, ~~from~~ 5% to 50% of solvents and ~~from~~ 0.5% to 6% of photoinitiators.

[0016]. Preferably, the transparent resins are selected
 5 from the group constituted by ~~aromatic or aliphatic~~ acrylic, polyester acrylate, urethane acrylate aromatic or aliphatic resins or mixtures thereof. Furthermore, the resins which have ~~shown themselves~~ proved to be particularly suitable ~~have been~~ were the resins with
 10 carboxylic functionality from 1 to 6 (declared and calculated according to the manufacturer) and, more preferably, the resins with functionality comprised of between 2 and 6.

[0017]. The monomers and methacrylic oligoethers are
 15 selected from the group ~~constituted by~~ consisting of Isobornyl methacrylate (IBOMA), Tetraethyleneglycol dimethacrylate (TTEGDMA), whilst the oligoethers and acrylate monomers are selected from Isobornyl acrylate (IBOA), 1,6 Hexanediol diacrylate (HDDA),
 20 Trimethylolpropane triacrylate (TMPTA), Tris(2-Hydroxyethyl) isocyanurate triacrylate (THEICTA), Tricyclodecane dimethanol diacrylate (TCDDMDA). Preferably, said monomers have functionality comprised of between 1 and 5. More preferably, said monomers and
 25 oligoethers are represented by Tris(2-Hydroxyethyl)

isocyanurate triacrylate (THEICTA), Tricyclodecane dimethanol diacrylate (TCDDMDA) and, in addition N-vinyl-2-pyrrolidone.

[0018]. The solvents used in the composition ~~in the~~
5 ~~subject~~therein are represented by the standard solvents normally added to varnishes such as, for example, esters, ketones or aromatic hydrocarbons. Amongst the esters methyl acetate and ethyl acetate can be used, amongst the ketones acetone and methyl ethyl ketone can be used,
10 whilst amongst the aromatic hydrocarbons toluol can be used.

[0019]. The photoinitiators, as mentioned ~~previously~~above, are substances capable of initiating the polymerisation process of the chemical composition when
15 they are subjected to light radiation. In particular, the photoinitiators according to the invention can be activated by various light sources such as solar or artificial light, preferably UV irradiation. In particular, the products which ~~have—shown~~
20 ~~themselves~~proved to be the best have been those which initiate polymerisation following exposure to UV-A type irradiation, or rather benzophenone derivatives amongst which the preferred are 4 methylbenzophenone and 2,4,6-trimethylbenzophenone, ketone derivatives, amongst which
25 are 1-hydroxy-cyclohexyl-phenyl-ketone, 2,2-dimethoxy-

1,2-diphenylethan-1-one, 2-hydroxy-2-methyl-1-phenyl-propan-1-one, methyl esters amongst which are the methyl ester of phenyl glyoxylic acid and phosphinoxides amongst which are the oxide of 2,4,6-trimethylbenzoyl-diphenyl-phosphine, or mixtures thereof.

[0020]. The above mentioned chemical composition can furthermore comprise additives, of the types widely known in the sector, having antioxidant functions which confer heat stability, such as for example the product IRGANOX 1010 pentaerythritol (tetrakis(3-(3,5-di-tert-butyl-4-hydroxyphenyl) propionate)) ~~as~~ sold by CIBA. A product which performs a protective role on the "clear coat" composition when it is exposed to sunlight can also be used. Amongst the products available on the market, in particular the derivative of hydroxyphenyltriazine into 1-methoxy-2-propanol TINUVIN 400, as sold by CIBA can be used. Moreover, a product which confers a hydrorepellant effect which increases surface slipperiness can be used. Chemically, this product is a silicone graft polymer which polymerises together with the other resins of the composition through radiation. An example of said product available on the market is BYK UV 3500 as sold by BYK.

[0021]. The additives can be present in quantities amounts ranging from 0% to 10%, preferably from 0.1% to 3%.

[0022]. The chemical composition just described can be produced according to a method which in general envisages the following sequential stages in sequence of:

- a) ~~the preparation of~~ preparing a solution comprising at least one resin amongst these ~~previously-listed~~ above and a suitable solvent or monomer;
- b) ~~the slowly~~ stirring ~~of~~ the solution for a time comprised of between 1' and 20' at a temperature comprised of between 58°C and 70°C;
- 10 c) ~~the addition~~ adding to the solution ~~of~~ at least one photoinitiator from amongst ~~those~~ previously-listed above following solubilisation in a suitable solvent;
- d) mixing for a time comprised of between 5 and 20 minutes so as to obtain a homogeneous solution and taking
- 15 care to keep the solution protected from solar or artificial light.

[0023]. In particular, stage a) is prepared with a solution comprising ~~from~~ 10 to 60 parts resins and ~~from~~ 40 to 90 parts solvents or monomers. When more resins are

20 used, the solution can be prepared by mixing ~~from~~ 10 to 70 parts resins with 30 - 80 parts monomers or solvents, heating to 58 - 70°C, with slow stirring, until obtaining a transparent solution, in a container made from AISI 304.

25 [0024]. In stage c), the photoinitiators, previously

solubilised in solvents at 40 - 60%, are added to the solution to a final percentage of ~~from~~ 1 - 6%.

[0025]. ~~In agreement with~~ According to an additional aspect of the present invention, the above chemical
5 composition can be polymerised according to a method which comprises exposure of the composition to UV type radiation.

[0026]. In fact, it has been surprisingly ~~been found~~ that if the subject chemical composition ~~in the subject~~
10 ~~is~~ is subjected to UV radiation, ~~its~~ the polymerisation process thereof is speeded up incredibly.

[0027]. The polymerisation by UV can be ~~realised~~ made by using various types of lamps such as high power and high pressure lamps which emit UV radiation in the B and
15 part of the C ranges. However, these types of lamps are dangerous and their use requires an environment ~~able~~ suitable to safeguard the operator from the dangerous types of radiation. Preferably, therefore, lamps which emit type A UV radiation and, more precisely, lamps which
20 emit radiation between 280 and 450 nm are used, which are extremely ~~more safely~~ safer than ~~those previous~~ above. The lamps are of the fluorescence- or metallic iodide-~~type in~~
~~type~~. Alternatively, lamps which also emit in the UV-B and UV-C regions can be used, ~~however but,~~ fitted provided
25 with special filters for the neutralisation of the B and

C type radiation.

- [0028]. In general, for static applications, ~~or~~
~~rather i.e.~~ when the lamp is ~~fixed~~ attached onto a
suitable support, lamps with power ratings comprised of
5 between 100W and 5KW or multi-lamp systems with power
outputs varying between 200W up to 5KW per lamp unit can
be used, even using more units. More particularly, the
lamp power outputs can vary from $0.1\text{W}/\text{cm}^2$ to $20\text{W}/\text{cm}^2$ of
irradiated surface.
- 10 [0029]. The exposure times ~~are dependent~~ on the "clear
coat" thicknesses applied, ~~on~~ the distance and the power
output of the lamp used. In general, exposures vary from
5 seconds to 15 minutes. For dry thicknesses of around 40
- 60 microns normally applied in car bodywork repair
15 operations, the exposure times can be considered to be
independent from the thicknesses. Since the UV
photopolymerisation is a function of the power output per
unit of surface area W/cm^2 , by increasing the distance of
the lamp, the lamp floodlight will irradiate a greater
20 surface area reducing the power in Watts per unit of
surface area. For example, by placing a 400 Watt lamp at
a distance of between 15 and 20 cm from the surface to be
irradiated, around 600 cm^2 will be irradiated with a
power of approx. $0.7\text{ W}/\text{cm}^2$ for such a period of time
25 necessary as to obtain good polymerisation comprised of

between 2 and 4 minutes. By using a lamp with a power output of 5,000 Watts placed at a distance of between 20 - 25 cm from the surface to be irradiated, 1,200 - 1,500 cm² are irradiated with a power output respectively comprised of between 4 W/cm² and 3 W/cm² and a time comprised of between 50 and 30 seconds. Such times are obviously inversely proportional to the power output per unit surface area irradiated by the UV lamp.

[0030]. For small repairs, ~~or i.e.~~ repairs which affect a surface area of from 600cm² to 6,000 cm², the lamp(s) are positioned on an appropriate fixed support, of a known type, which can allow the rapid positioning at the level of the surface ~~for to be treated~~ment. In the case in which the polymerisation must affect a more enlarged surface or even the entire vehicle bodywork, it is possible to mount the lamps onto robotised rigs which are able to homogeneously irradiate complex profile surfaces ~~in a homogeneous manner~~ by performing a scan of the surface to be irradiated with a scanning speed directly proportional to the power output of the UV system used. In this case, it is preferable to use lamps with power outputs varying between 2.5KW to 25KW.

[0031]. ~~According to~~In agreement with an additional further subject aim of the invention, in the following will be described a treatment method for vehicle bodywork

surfaces or parts ~~of them~~thereof with the above mentioned ("clear coat") chemical composition.

[0032]. The treatment method comprises the following sequential stages in sequence of:

- 5 i) ~~the making available~~providing of ~~aa~~ polymerisable chemical composition comprising ~~from~~ 10% to 60% of transparent hydroxylated acrylic resins, ~~from~~ 10% to 70% of monomers selected from oligoethers and acrylate or methacrylate monomers, ~~from~~ 0% to 90% of solvents and
10 from 0.1% to 10 % of photoinitiators;
- ii) ~~the application of~~applying a layer of said chemical composition onto the surface to be treated;
- iii) leaving the solvent contained in said layer of said chemical composition to evaporate;
- 15 iv) irradiating said layer with a UV irradiation lamp for ~~such a~~ time sufficient as to ~~essentially~~ substantially obtain ~~its~~ the complete polymerisation thereof.

[0033]. The polymerisable chemical composition is preferably the ~~previously detailed~~ composition described
20 above.

[0034]. The application stage ii) of a layer of said composition preferably occurs through the deposition in the form of a film previously diluted with an appropriate solvent, such as these previously described. The film
25 which is deposited has a thickness which can vary from 10

microns to 100 microns and its viscosity can vary from 12 to 18 seconds in a Ford #4 cup (according to the ASTM system) and according to the degree of dilution adopted.

[0035]. The evaporation stage iii) of the solvent will
 5 depend on the conditions adopted from time to time ~~or~~
~~rather on~~i.e. the thickness of the deposited film, ~~on~~ the
 amount of solvent used and ~~on~~ the chemical composition
 used. In any case, in general, the evaporation time
 ("flash off") varies from 1 minute to 5 minutes.

10 [0036]. The irradiation stage iv) of said layer can
~~take place in agreement with~~occur according to the
 previously described polymerisation ~~procedure~~process.

[0037]. In addition, the treatment method comprises the
 following steps prior to the application of the
 15 polymerisable chemical composition:

- the reshaping or ~~substitution~~replacement of the
 damaged bodywork parts;
- the filling and sandpapering of the surface of said
 damaged parts;
- 20 - the application of one or more ~~coats~~ of
~~foundation~~primers onto said surface and the relative
 polymerisation;
- the preparation of said surface for the application of
 the "base coat" using abrasive papers;
- 25 - the application of the "base coat".

[0038]. The steps just described are entirely conventional in the vehicle bodywork repair sector whereby they will not be described in any further detail.

[0039]. Finally, the treatment method can ~~envisage~~
 5 provide a last stage consisting of finishing, ~~or~~
~~rather~~ i.e. the elimination of any possible defects arising during the execution of the preceding stages, in agreement with known techniques.

[0040]. From ~~that~~ what presented stated up to now, the
 10 numerous advantages brought about by the polymerisable chemical composition ~~are clear, from its~~ the polymerisation procedure thereof as well as the treatment method for the bodywork of vehicles or parts ~~of them~~ thereof with said composition are evident.

15 [0041]. Firstly, the chemical composition of the invention does not require any catalyst ~~in order~~ to activate the polymerisation process. This means that even the loss of time, due to the time required for the mixing of the two components, necessary in the case of the use
 20 of the above mentioned "clear coat 2K", is avoided. ~~As a consequence~~ Consequently, the preparation of the polymerisable chemical composition according to the invention is simpler to carry out and allows a substantial saving in production costs.

25 [0042]. Secondly, the polymerisation stage of the

polymerisable chemical composition ("clear coat") on average is completed within 5-7 minutes, whilst the ~~later~~ subsequent ~~possible~~ optional finishing stage can be accomplished after a few minutes. On the contrary, as
5 previously reported, according to the known art, said stage requires oven cooking times comprised of between 30 and 60 minutes or 24 hours at room temperature. Therefore, the time saving which the composition of the invention brings about is incredible.

10 [0043]. In addition, the above mentioned polymerisation is very rapid and the ~~realisation~~ manufacturing costs are very low with respect to these which must be faced with the oven cooking method, as explained in the introductory section of the present description. In fact, the use of
15 UV lamps for a few minutes for the polymerisation is much more economical and faster ~~with respect to~~ than the use of high temperature ovens which work through the combustion of very expensive fuels.

[0044]. Surprisingly, it has also been observed that
20 the polymerisation with UV ~~allows~~ enables the attainment of polymerisation percentages close to 100% in very short times, as will be apparent from the comparative test reported ~~herein after~~ below. From that it arises that even the finishing can be ~~realised~~ made with great ease and
25 rapidity and however with optimal results.

[0045]. In the following are reported some example embodiments, given as non limiting indications of the invention.

EXAMPLE 1

5 Preparation of the (transparent) "clear coat"
polymerisable chemical composition

[0046]. 12.32 g of tris (2-hydroxyethyl) isocyanurate triacrylate, produced by Cray Valley and sold under the commercial name Sartomer SR 368, and 35.46 g of methanol
10 diacrylate tricyclodecane, produced by Cray Valley and marketed under the name of Sartomer SR 833S are weighed in an AISI 304 steel basin and heated to 58°C. They are slowly stirred until complete solution is obtained. Afterwards, 25.46 g of hydroxylated acrylic polyol with
15 1-2% of OH with reference to the dry weight of the resin are added. This product is available on the market from various manufacturers, as 50% of the dry resin in butyl acetate, such as Desmophen A450 from Bayer, Setalux 1184 SS-51 from AKZO RESINS, Domacryl 546 from HELIOS. In the
20 present example Setalux 1184 SS-51 has been used. 14.01 g of N-vinyl-2-pyrrolidone produced by BASF are also added along with the additives TINUVIN 400 at 50% in ethyl acetate in an amount of 0.6 g, IRGANOX 1010 at 20% in ethyl acetate in an amount of 1.5 g and BYK UV 3500 in an
25 amount of 0.4 g. Slow stirring takes place for 15

minutes. Later, the following amounts of photoinitiators, previously solubilised in 6 g of cyclohexanone, are added:

- 0.7 g of IRGACURE 184;
- 5 - 3.2 g of DAROCUR TPO;
- 0.35 g of DAROCUR MBF.

The resulting product is mixed for 10 minutes so as to obtain homogeneity of the composition, taking care to keep the product from exposure to solar or artificial
10 light thereby avoiding the danger of polymerisation. At this point the polymerisable chemical composition is stored in screw topped metallic containers and maintained at room temperature.

EXAMPLE 2

15 Polymerisation procedure of the polymerisable chemical composition

[0047]. A steel panel according to European standard ISO 1514 has been employed as a sample in order to evaluate the polymerisation of the composition of the
20 invention. Onto the surface of said panel has been applied an acrylic ~~foundation~~ primer constituted by "Acrylic ~~Foundation~~ Primer F51 2K" catalysed in the ratio of 5:1 with "C16 Universal Catalyst" both produced by ICR Spa. Following 24 hours of polymerisation at room
25 temperature, the panel has been dry sanded with abrasive

papers of decreasing grain from P150 to P600, manufactured by 3M. Afterwards, the surface of the panel thus treated has been covered with a 20 micron layer of BASISLACK type metalicised "base coat" as sold by
5 STANDOX, diluted to 60% with "11040" the specific diluent for metalicised paints. The "base coat" has been left to air dry at room temperature for 3 minutes. At this point, onto the dried "base coat" has been applied a layer of approx. 50 microns of the composition of example 1
10 ("clear coat") following dilution with ethyl acetate so as to obtain a viscosity of 15" in a Ford #4 cup. After three minutes from said application, so as to allow the evaporation of the solvent, the panel has been irradiated for 2 minutes with an OSRAM Ultramed lamp with a power
15 output of 400W placed at a distance of 20 cm from the surface treated with the above mentioned composition.

EXAMPLE 3

Comparative application test of the polymerisable chemical composition according to the invention with
20 respect to a composition of the known art

[0048]. In order to test the effectiveness of the polymerisable chemical composition of example 1, six steel panels according to the European standard ISO 1514 have been prepared. A ~~foundation~~ primer as described in
25 example 2 has been applied onto all the panels and the

same procedure has been followed up to the application of the "base coat". Afterwards, the test surfaces on 4 of the six panels have been covered with approx. 50 microns of "H61 Transparent Acrylic 2K" type "clear coat 2K" as sold by ICR Spa, catalysed by an aliphatic isocyanate based catalyst "Universal Catalyst C15" as produced by ICR Spa, in a ratio of 2:1 and diluted with 25% of polyurethane diluent "Diluent D10" (again from ICR Spa) so as to obtain an application viscosity of 18'' in a Ford #4 cup. Two panels have been subjected to a cooking cycle in an oven for 30 minutes at 60°C, whilst another two have been kept at a temperature of around 20°C for a total time of 7 days. Instead, the remaining two panels have been treated in agreement with example 1 and example 2.

[0049]. Hardness tests carried out 10 minutes after the end of the polymerisation cycle, by leaving the panels at room temperature have given the following results reported in table I. For the hardness tests, the Persoz method has been used at room temperature.

TABLE I

Product	Polymerisation time (minutes /hours/ days)	Polymerisation temperature (°C)	Resting time following polymerisation (minutes/hours/days)	hardness
Clear coat 2k	30'	60°	10 minutes	180''

Clear coat 2k	30'	60°	4 hours	190''
Clear coat 2k	30'	60°	24 hours	285''
Clear coat 2k	30'	60°	7 days	330''
Clear coat 2k	4 hours	20°	-	70''
Clear coat 2k	24 hours	20°	-	245''
Clear coat 2k	7 days	20°		320''
Clear coat UV	2' under the conditions of example 1	-	10'	340''
Clear coat UV	2' under the conditions of example 1	-	24 hours	352''
Clear coat UV	2' under the conditions of example 1	-	7 days	355''

[0050]. The hardness values reported in table I clearly indicate that finishing operations such as polishing can be quickly carried out on the "clear coat UV" layer applied ~~in agreement with~~according to the present invention, whilst that is not possible in the case ~~of the use of the~~ "clear coat 2K" of the known art is used since the surface is not sufficiently hardened. In fact, sufficient hardness in order to allow such operations with the known art is obtained only after at least 4

5

10 hours from the end of the oven cooking cycle at 60°C. In

any case, it is to be noted that optimal hardness (280-300'' Persoz) is obtained only after at least 24 hours.

EXAMPLE 4

Adhesion test

5 [0051]. On the panels subjected to the Persoz hardness test, adhesion tests have been carried out according to European standard ISO 4624. The adhesion of the chemical composition of the invention onto the "base coat" has been close to 100%.

10 EXAMPLE 5

Application test

[0052]. The polymerisable chemical composition of example 1 has been applied onto parts of the bodywork of vehicles in order to evaluate the application
15 characteristics. The applications have affected different sections of the bodywork over areas of approx. 20x20 cm and in every case a surface deformation has been ~~provoked~~ caused by hitting the bodywork with a hammer. Later, the deformed part has been filled with "S01 UV Monocomponent
20 polyester filler" filler ~~as~~ produced by ICR Spa and sanded with P150 abrasive paper from 3M. Later, an "F01 Filling ~~Foundation~~ Primer UV" type ~~foundation~~ primer ~~as~~ produced by ICR Spa has been applied. The application of the "base coat" was then performed as ~~previously~~
25 described above. Afterwards, the procedure of example 1

has been carried out by first applying a 70 - 80 micron layer of the chemical composition diluted to 50% with ethyl acetate. The same chemical composition has then been diluted again with 25% ethyl acetate and has been applied by spraying, with a specific conventional spray gun, taking care to widen out over the original varnish. Following three minutes of rest in order to allow for the evaporation of the solvent, the surface thus treated has been irradiated for 2 minutes under the conditions described in example 3. Once the treated bodywork part has reached room temperature (approx. 10 minutes), the surface has been blended in with the non treated surface by finishing (polishing) with abrasive (cutting) paste and polish.

[0053]. The various tests carried out on the different "base coats", even with very clear or dark colours, have given a very valid result to ~~such a point~~such an extent that the treated parts were no longer distinguishable from the originals.

[0054]. In addition, tests have also been carried out on surfaces which had not been prepared with abrasives, or polished. Nevertheless, optimal adhesion in obtaining approx. 100% has been found.

[0055]. The polymerisable chemical composition, the relevant polymerisation procedure as well as the

treatment method for bodywork or parts of them allow
therefore the resolution of the inconveniences caused by
the use of the compositions and the methods of the known
art, described in the introductory section of the present
5 description, and provide numerous, already described,
advantages.

[0056]. Variant embodiments of the polymerisable
chemical composition, of the relevant polymerisation
procedure as well as the treatment method of bodywork or
10 parts of them ~~in agreement with~~ according to the invention
are within the range of the expert in the sector and
however fall within the ~~safeguarding~~ protection scope
~~environment~~ of the following claims.